

	ICS Growstar Trolessional	Revision: 11/01/2016
	(20-0-10)	Supersedes Revision: 07/21/2016
	1. Product and Company Id	entification
Product Code:	902467	
Product Name:	TCS Growstar Professional Turf Fertilizer	r (20-0-10)
Trade Name:	Granular Fertilizer	
Company Name:	Turf Care Supply Corp.	Phone Number:
	50 Pearl Road Suite 200	1 (330)558-0910
	Brunswick, OH 44212	
Web site address:	www.turfcaresupply.com	
Email address:	regaffairs@tcscusa.com	
Emergency Contact:	PERS	1 (800)633-8253
Information:	Turf Care Supply Corp.	1 (330)558-0910
Synonyms:	Granular Fertilizer	
	2. Hazards Identifica	tion
Acute Toxicity: Oral, Catego		
GHS Signal Word:	Warning	
GHS Hazard Phrases:	Harmful if swallowed.	
		ye irritation. May cause repiratory irritation.
	exposure.	n and lungs through prolonged or repeated
GHS Precaution Phrases:	Avoid breathing dust.	
	Wear protective gloves, protective clothir	ng, and eye protection.
	Call a POISON CENTER or doctor/physi	cian if you feel unwell.
GHS Response Phrases:	If eye irritation persists, get medical advid	ce/attention.
		for several minutes. Remove contact lenses, if
	present and easy to do so. Continue rins	•
GHS Storage and Disposal		vent uncontrolled release to the environment.
Phrases:	Store in a closed container.	cording to lobal directions, dispass of container
	If material cannot be completely used ac and contents according to section 13.	cording to label directions, dispose of container
Detential Linelth Effects	-	not may asy a dormatitic. Declar and an
Potential Health Effects	Chronic: Prolonged or repeated skin cont	act may cause dermatitis. Prolonged of

(Acute and Chronic): repeated exposure may cause permanent eye damage. Chronic exposure may cause lung damage. Effects may be delayed.

Inhalation:May be harmful if inhaled. Low hazard for normal industrial handling. The toxicological<br/>properties of this substance have not been fully investigated. May cause systemic<br/>effects. Material may be irritating to mucous membranes and upper respiratory tract.Skin Contact:May cause skin irritation. Dust causes mechanical irritation. Low hazard for usual

Skin Contact:May cause skin irritation. Dust causes mechanical irritation. Low hazard for usual<br/>industrial handling.Eye Contact:May cause eye irritation. Dust may cause mechanical irritation.

Ingestion: May be harmful if swallowed. May cause gastrointestinal irritation with nausea, vomiting and diarrhea. Low hazard for normal industrial handling. The toxicological properties of this substance have not been fully investigated. May cause systemic effects.

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	3	. Composition/Inf	formation on Ingred	lients	
CAS #	Hazardous Com	ponents (Chemical Name)	Concentration		
57-13-6	Urea		43.5 %		
1317-65-3	Limestone		22.0 - 22.7 %		
7447-40-7	Potassium chlori	de	15.9 %		
8047-67-4	Saccharated iron	oxide	6.80 %		
1344-43-0	Manganese oxid	e	2.58 - 2.73 %		
1309-48-4	Magnesium oxide	e (MgO)	1.66 %		
112926-00-8	Precipitated silicated	a	0.886 - 1.17 %		
14808-60-7	Quartz		0.904 - 0.961 %		
		4. First	Aid Measures		
Emergency a	nd First Aid				
Procedures:					
In Case of In	halation:	-	and move to fresh air immed is difficult, give oxygen. Get n	iately. If not breathing, give artificial nedical aid.	
In Case of Sk	in Contact:	Get medical aid if irritation develops or persists. In case of contact, flush skin with plenty of water. Remove contaminated clothing and shoes. Get medical aid if irritation develops and persists. Wash clothing before reuse. Wash off with soap and plenty of water.			
In Case of Eye Contact:		Flush eyes with plenty of water for at least 15 minutes, occasionally lifting the upper and lower eyelids. Get medical aid. Do NOT allow victim to rub eyes or keep eyes closed.			
		Get medical aid. If victim is conscious and alert, give 2-4 cupfuls of milk or water. Call a poison control center. If swallowed, do NOT induce vomiting unless directed to do so by medical personnel. Never give anything by mouth to an unconscious person.			
		To the best of our knowled not been thoroughly inve		, and toxicological properties have	
Note to Phys	ician:	Treat symptomatically ar	nd supportively.		
		5. Fire Fig	hting Measures		
Flash Pt:		No data.			
Explosive Lir	nits:	LEL: No data.	UEL: No data.		
Autoignition		No data.			
U			chemical, carbon dioxide, or v e, alcohol-resistant foam, or v	water spray. For large fires, use dry vater spray.	
MSHA/t noncom		MSHA/NIOSH (approved noncombustible. Decom	As in any fire, wear a self-contained breathing apparatus in pressure-demand, MSHA/NIOSH (approved or equivalent), and full protective gear. Substance is noncombustible. Decomposes at high temperatures, resulting in toxic and corrosive products. Runoff from fire control or dilution water may cause pollution.		
•		•	of this product are non-comb at elevated temperatures.	oustible. However, a portion of them	
Hazardous C Products:	ombustion	chlorine, cyanic acid, an potassium, sulfur, and cl metals used as nutrients	d cyanide, and oxides of carb hlorine, and oxides of alkaline	of ammonia, formaldehyde, biuret, bon, nitrogen, phosphorus, e earth metals, and certain heavier s copper, iron, manganese, and	

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	6. Accidental Release Measures
Steps To Be Taken In Case Material Is Released Or Spilled:	Use proper personal protective equipment as indicated in Section 8. Spills/Leaks: Vacuum or sweep up material and place into a suitable disposal contain Avoid generating dusty conditions. Provide ventilation. Avoid runoff into storm sewers and ditches which lead to waterways. Do not let this product enter the environment except as directed on product label. Clean up spills immediately, observing precaution in the Protective Equipment section.
	Personal precautions. Use personal protective equipment. Avoid dust formation. Avoid breathing dust. Ensu adequate ventilation.
	Environmental precautions. Do not let product enter drains.
	Pick up and arrange disposal without creating dust. Keep in suitable, closed container for disposal.
	PROCEDURES & PERSONAL PRECAUTIONS. Exercise appropriate precautions to minimize direct contact with skin or eyes and prevent inhalation of dust.
	Methods for cleaning up. Sweep up, place in a bag and hold for waste disposal. Avoid raising dust. Ventilate ar and wash spill site after material pickup is complete.
	7. Handling and Storage
Precautions To Be Taken in Handling:	Use with adequate ventilation. Minimize dust generation and accumulation. Avoid contact with eyes, skin, and clothing. Avoid ingestion and inhalation. Wash thoroughly after handling. Use only in a well-ventilated area. Keep container tightly closed. Wash clothing before reuse.
	Provide appropriate exhaust ventilation at places where dust is formed.
Precautions To Be Taken in Storing:	Store in a cool, dry place. Keep container closed when not in use.
8	3. Exposure Controls/Personal Protection
CAS # Partial Chemical	

CAS #	Partial Chemical Name	OSHA TWA	ACGIH TWA	Other Limits	
57-13-6	Urea	No data.	No data.	No data.	
1317-65-3	Limestone	PEL: 15 (dust); 5 (resp.) mg/m3	No data.	No data.	
7447-40-7	Potassium chloride	No data.	No data.	No data.	
8047-67-4	Saccharated iron oxide	No data.	No data.	No data.	
1344-43-0	Manganese oxide	CEIL: 5 mg/m3	TWA: 0.02 mg/m3 (resp.) 0.1 mg/m3 (IHL)	No data.	
1309-48-4	Magnesium oxide (MgO)	PEL: 15 (particulate) mg/m3	TLV: 10 mg/m3 (Inhalation)	No data.	
112926-00-8	Precipitated silica	PEL: 80 mg/m3/(%SiO2)	TLV: 10 mg/m3	No data.	
14808-60-7	Quartz	PEL: 50 ug/m3	TLV: 0.05 mg/m3 (R)	No data.	



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Respiratory Equipment (Specify Type):	A respiratory protection program that meets OSHA's requirements or European Standard EN 149 must be conditions warrant respirator use. Where protection f desired, use type N95 (US) or type P1 (EN 143) dust use type OV/AG/P99 (US) or type ABEK-P2 (EU EN	e followed whenever workplace from nuisance levels of dusts are t masks. For higher level protection
Eye Protection:	Wear appropriate protective eyeglasses or chemical OSHA's eye and face protection regulations in 29 CF EN166.	
Protective Gloves:	Wear appropriate protective gloves to prevent skin ex	xposure. Wash and dry hands.
Other Protective Clothing:	Wear appropriate protective clothing to prevent skin e according to the amount and concentration of the dat place.	
Engineering Controls (Ventilation etc.):	Facilities storing or utilizing this material should be ea a safety shower. Use adequate ventilation to keep air adequate general or local exhaust ventilation to keep permissible exposure limits.	rborne concentrations low. Use
Work/Hygienic/Maintenance Practices:	Handle in accordance with good industrial hygiene an before breaks and at the end of workday. Wash thore	• •
	9. Physical and Chemical Properti	es
Physical States:	[]Gas []Liquid [X]Solid	
Appearance and Odor:	Multi-colored, granular solid. Slight ammonia-like odor.	
pH:	No data.	
Melting Point:	~ 133 C	
Boiling Point:	No data.	
Flash Pt:	No data.	
Evaporation Rate:	No data.	
Flammability (solid, gas):	No data available.	
Explosive Limits:	LEL: No data. UEL: No data.	
Vapor Pressure (vs. Air or mm Hg):	No data.	
Vapor Density (vs. Air = 1):	No data.	
Specific Gravity (Water = 1):		
Bulk density:	~ 45 - 65 LB/CF	
Solubility in Water:	~ 1,080 g/l at 20.0 C	
Solubility Notes:	The solubility value cited is for the urea component o section 3.	if this product, if present. See
Octanol/Water Partition Coefficient:	No data.	
Autoignition Pt:	No data.	
Decomposition Temperature	: ~ 135 C	
Viscosity:	No data.	
Additional Physical Information	The melting point and decomposition temperatures c this product, if present. See section 3. Urea decomposes before boiling. (UNEP Publication 57-13-6)	
	57-13-6)	



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10. Stability and Reactivity				
Unstable [ ] Stable [ X ]				
Incompatible materials, dust generation, heating to decomposition. High temperatures.				
<b>b</b> Strong oxidizing agents, bases, acids, aluminum.				
<b>r</b> The decomposition of fertilizer products may result in the generation of some or all of the following: ammonia, formaldehyde, biuret, chlorine, cyanic acid, and cyanide, and oxides of carbon, nitrogen, phosphorus, potassium, sulfur, and chlorine, and oxides of alkaline earth metals, and certain heavier metals used as nutrients in fertilizer products, such as copper, iron, manganese, and zinc, and other irritating and toxic fumes and gases.				
Will occur [ ] Will not occur [ X ]				
No data available.				
11. Toxicological Information				
Epidemiology: No information found. Teratogenicity: Teratogenic effects have occurred in experimental animals. Neurotoxic effects have occurred in experimental animals. Reproductive toxicity - no data available. Inhalation: May cause damage to organs through prolonged or repeated exposure.				
CAS# 57-13-6: Urea: Other Studies:, TCLo, Inhalation, Rat, 288.0 MG/M3, 17 W; Gigiena Truda i Professional'nye Zabolevaniya.(Labor Hygiene and Occupational Disease), V/O Mezhdunarodnaya Kniga, Moscow 113095 Russia, Vol/p/yr: 30(3),43, 1986				
Acute toxicity, LD50, Oral, Rat, 8471. MG/KG; Gigiena i Sanitariya, Mezhdunarodnaya Kniga, ul. B. Yakimanka, 39, 113095, Moscow 113095 Russia, Vol/p/yr: 51(6),8, 1986				
Standard Draize Test, Skin, Human, 22.00 MG, 3 D; Cutaneous Toxicity, Proceedings o the 3rd Conference, 1976, D, V.A., and P. L, New York, Academic Press, Inc., London United Kingdom, Vol/p/yr: -,127, 1977				
CAS# 7447-40-7: Potassium chloride: Acute toxicity, LD50, Oral, Rat, 2600. MG/KG; "Sbornik Vysledku Toxixologickeho Vysetreni Latek A Pripravku,", Institut Pro Vychovu Vedoucicn P, Marhold, J.V., Institut Pro Vychovu Vedoucicn, Pracovniku Chemickeho, Prumyclu Praha Czechoslovakia, Vol/p/yr: -,8, 1972				
Standard Draize Test, Eyes, Species: Rabbit, 500.0 MG, 24 H; "Sbornik Vysledku Toxixologickeho Vysetreni Latek A Pripravku,", Institut Pro Vychovu Vedoucicn P, Marhold, J.V., Institut Pro Vychovu Vedoucicn, Pracovniku Chemickeho, Prumyclu Prah Czechoslovakia, Vol/p/yr: -,8, 1972				
This material may contain small amounts of respirable crystalline and amorphous silica. The International Agency for Cancer Research (IARC) has classified crystalline silica as				

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carcinogenicity to humans (Group 3). See "Silica, Some Silicates, Coal dust and
para-Aramid Fibrils in IARC Monographs on the Evaluation of Carcinogenic Risks to
Humans", (Vol. 68).

CAS #	Hazardous Components (Chemical Name)	NTP	IARC	ACGIH	OSHA
57-13-6	Urea	n.a.	n.a.	n.a.	n.a.
1317-65-3	Limestone	n.a.	n.a.	n.a.	n.a.
7447-40-7	Potassium chloride	n.a.	n.a.	n.a.	n.a.
8047-67-4	Saccharated iron oxide	n.a.	3	n.a.	n.a.
1344-43-0	Manganese oxide	n.a.	n.a.	n.a.	n.a.
1309-48-4	Magnesium oxide (MgO)	n.a.	n.a.	A4	n.a.
112926-00-8	Precipitated silica	n.a.	n.a.	n.a.	n.a.
14808-60-7	Quartz	Known	1	A2	n.a.

**12. Ecological Information** 

General Ecological

Environmental: If released to the atmosphere, urea will degrade rapidly in the vapor-phase by reaction with photochemically produced hydroxyl radicals (half-life of 9.6 hr). If released to soil, urea is hydrolyzed to ammonium through soil urease activity (the basis of its use as a fertilizer). The rate of hydrolysis can be fast (24 hr); however, a number a variables (such as increasing the pellet size of the fertilizer) can decrease the degradation rate from days to weeks.

Other: Do not empty into drains.

Other: Estimated BCF value = 0.05. This value indicates that this product will exhibit low bioconcentration in aquatic organisms. Biodegradation is expected to be an important fate process in water. It has a low potential to affect aquatic systems. If diluted with water, this chemical released directly or indirectly into the environment is not expected to have a significant impact.

CAS# 57-13-6: Urea:

Lethal concentration to 0% of test organisms., Creek Chub (Semotilus atromaculatus), 16000000. UG/L, 24 H, Mortality, Water temperature: 15.0 C - 21.0 C C, pH: 8.30, Hardness: 98.00 MG/L; Appraisal of a Chemical Waste Problem by Fish Toxicity Tests, Gillette, L.A., D.L. Miller, and H.E. Redman, 1952

 CAS# 7447-40-7: Potassium chloride: LC50, Rainbow Trout (Oncorhynchus mykiss), 1610000. UG/L, 48 H, Mortality, Water temperature: 17.0 C C, pH: 7.70, Hardness: 40.00 MG/L; Toxicity of Candidate Molluscicides to Zebra Mussels (Dreissena polymorpha) and Selected Nontarget Organisms, Waller, D.L., J.J. Rach, W.G. Cope, L.L. Marking, S.W. Fisher, and H. Dabrowska, 1993
Persistence and Degradability:

Bioaccumulative Potential:No data available.Mobility in Soil:No data available.



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		12_D;	`	,		ersedes Revision: 07/21/2016	
			-	Consideratio			
Vaste Dispos	al Method:	and contents ac		-	ng to label direction	ons, dispose of containe	
		Contact a licens	ed professio	onal waste disposa	al service to dispo	se of this material.	
		Do not let product enter drains.					
		as a hazardous in 40 CFR Parts	waste. US E 261. Additio		the classification erators must cons		
		RCRA P-Series RCRA U-Series					
		Observe all fede	eral, state, a	nd local environm	ental regulations.		
		14. ]	ranspol	rt Informatio	n		
	PORT (US DOT						
DOT Haza	rd Class:	me: Not Regulat					
UN/NA Nu	mber:						
		15. R	egulato	ry Informatio	on		
	-	nents and Reautho		-	0.00/00		
<b>CAS #</b> 57-13-6	Hazardous Con Urea	ponents (Chemica	al Name)	<b>S. 302 (EHS)</b> No	<b>S. 304 RQ</b> No	<b>S. 313 (TRI)</b> No	
1317-65-3	Limestone			No	No	No	
7447-40-7	Potassium chlori	de		No	No	No	
8047-67-4	Saccharated iror	n oxide		No	No	No	
1344-43-0	Manganese oxid	e		No	No	Yes-Cat. N450	
1309-48-4	Magnesium oxid			No	No	No	
112926-00-8	Precipitated silic			No	No	No	
14808-60-7	Quartz			No	No	No	
	ories' defined III Sections	[X] Yes [] No [X] Yes [] No [] Yes [X] No [] Yes [X] No [] Yes [X] No	Chronic (de Fire Hazar	elease of Pressure	zard		
CAS #	Hazardous Con	ponents (Chemica	al Name)	Other US EPA o	r State Lists		
57-13-6	Urea			Inventory, 8A CA		No; TSCA: Yes - No; MA Oil/HazMat: No; NY Part 597: No; PA HSL	
1317-65-3 7447-40-7	Limestone Potassium chlori	de		Inventory; CA P Part 5: No; NJ E		0il/HazMat: No; MI CMR, 97: No; PA HSL: Yes - 1	
				,	.,		



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		Inventory; CA PROP.65: No; MA Oil/HazMat: No; MI CMR,
		Part 5: No; NJ EHS: No; NY Part 597: No; PA HSL: No
8047-67-4	Saccharated iron oxide	
8047-67-4	Saccharated from oxide	CAA HAP,ODC: No; CWA NPDES: No; TSCA: No; CA
		PROP.65: No; MA Oil/HazMat: No; MI CMR, Part 5: No; NJ
		EHS: No; NY Part 597: No; PA HSL: No
1344-43-0	Manganese oxide	CAA HAP,ODC: Yes - Cat.; CWA NPDES: No; TSCA: Yes -
		Inventory; CA PROP.65: No; MA Oil/HazMat: No; MI CMR,
		Part 5: Yes - Cat.; NJ EHS: Yes - Cat.; NY Part 597: No; PA
		HSL: No
1309-48-4	Magnesium oxide (MgO)	CAA HAP, ODC: No; CWA NPDES: No; TSCA: Yes -
		Inventory; CA PROP.65: No; MA Oil/HazMat: No; MI CMR,
		Part 5: No; NJ EHS: No; NY Part 597: No; PA HSL: Yes - 1
112926-00-8	Precipitated silica	CAA HAP.ODC: No; CWA NPDES: No; TSCA: Yes -
		Inventory; CA PROP.65: No; MA Oil/HazMat: No; MI CMR,
		Part 5: No; NJ EHS: No; NY Part 597: No; PA HSL: Yes - 1
14808-60-7	Quartz	CAA HAP,ODC: No; CWA NPDES: No; TSCA: Yes -
14000-00-7	Qualiz	
		Inventory; CA PROP.65: No; MA Oil/HazMat: No; MI CMR,
		Part 5: No; NJ EHS: No; NY Part 597: No; PA HSL: Yes - 1
		Other Information
Revision Date	<b>e:</b> 11/01/2016	
		Elemmehility A. Instability

Hazard Rating System:



Additional Information About No data available.

This Product:

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